國立臺灣大學105學年度轉學生招生考試試題

題號: 50 科目:有機化學

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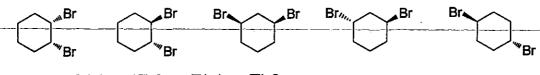
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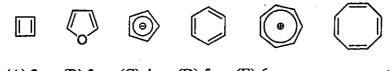
※注意:請於試卷上「選擇題作答區」依序作答。

第一部份: 單遷題 (3pts each)

- 1. Which of the following species can serve as a nucleophile?
 - (a) H⁺ (b) MeOH (c) BF₃ (d) Ph₃P (e) Et₃N
 - (A) abcde (B) bcde (C) ade (D) bde (E) bcd
- 2. Which of the following species can serve as a *Lewis acid*?
 - (a) Et₂O (b) AlBr₃ (c) BF₃ (d) FeCl₃ (e) Et₃N
 - (A) abcde (B) bcd (C) abc (D) bce (E) ace
- 3. Which of the following nitrogen-containing compound is *most basic*?
 - (A) acetamide (B) pyridine (C) triethylamine (D) aniline (E) pyrrole
- 4. How many of the following compound(s) can be deprotonated by lithium diisopropylamide (LDA)?
 - (a) 1-hexene (b) 1-hexyne (c) cyclohexanone (d) cyclohexanol (e) cyclohexene
 - (A) 1 (B) 2 (C) 3 (D) 4 (E) 5
- 5. How many of the following compounds are chiral?



- (A) 1 (B) 2 (C) 3 (D) 4
- 6. How many of the following species are aromatic?



(A) 2 (B) 3 (C) 4 (D) 5 (E) 6

(d) BH₃, THF, then H₂O₂/OH

- 7. Which of the reagents(s) listed can complete the following reaction?
 - (a) KMnO₄, heat (b) DMSO, (COCl)₂, -60°C, then Et₃N
 - (c) O₃, CH₂Cl₂, -78°C, then Me₂S (d) Jones reagent

(e) PCC, CH₂Cl₂
(A) abcde (B) abde (C) ade (D) be (E) ad

8. Which of the reagent(s) listed can **NOT** generate any alcohol-containing product from the following reaction?

(a) KMnO₄, heat (b) Br₂, H₂O (c) aqueous H₂SO₄

(A) abcde (B) bcde (C) ae (D) bcd (E) none, all generate alcohol

(e) MCPBA

見背面

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9. In a typical ¹³C NMR spectrum, how many ¹³C signals can you observe in the following molecule?

10. Which of the reagent can complete the following transformation?

- 11. Which of the following reaction(s) can afford a meso compound as the major product?
 - (a) Cyclohexene reacts with MCPBA.
 - (b) Cyclohexene reacts with Br2.
 - (c) Cyclohexene reacts with KMnO4 in a cold basic solution.
 - (d) Cyclohexene reacts with H2 in the presence of Pd catalyst.
 - (e) Cyclohexene reacts with aqueous H₂SO₄ solution.
 - (A) ac (B) abc (C) acd (D) bde (E) abcde
- 12. In the reaction between <u>sodium cyanide</u> and <u>(R)-2-bromopentane</u> in a certain <u>solvent</u>, which of the following statement is correct? (PS: [A] = concentration of A)
 - (A) The reaction is faster at 25°C than at 100°C.
 - (B) The reaction is independent of [sodium cyanide].
 - (C) The reaction is faster when [(R)-2-bromopentane] is increased.
 - (D) The major product also has the R configuration.
 - (E) The reaction is faster in MeOH solvent than in DMSO solvent.

※注意:請於試卷上「非選擇題作答區」標明題號並依序作答。

第二部份:問答題

13. Provide the reagent C, E, G and <u>major</u> product A, B, D, F to complete the following synthetic scheme. (3pts each)

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14. Rank the following compounds from the most acidic to the least acidic. (4pts)

15. Provide the *major* product A to F to complete the following synthetic scheme. (3pts each)

F:
$$C_7H_{12}O$$

(1) OH^-
(2) H^+ , heat

E: $C_{10}H_{16}O_3$

NaOEt

OEt

OEt

(1) NaOEt
(2) H^+
(2) H^+
(2) H^+
(2) H^+
(3) PBr₃

PBr₃

16. Provide the *major* product A to G to complete the following synthetic scheme. (3pts each)

(Hint: The NMR spetrums of coupound A are shown here)

¹H NMR (CDCl₃) δ (ppm): 7.378 (doublet, 2H), 7,050 (doublet, 2H), 2,311 (singlet, 3H)

¹³C NMR (CDCl₃) δ (ppm): 136.73, 131.22, 130.79, 119.05, 20.88 (five peaks)

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