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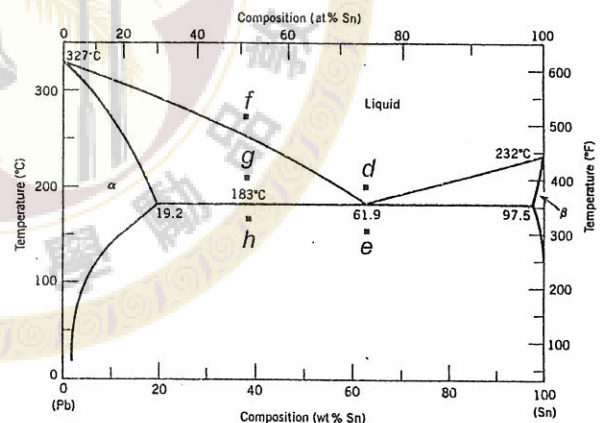
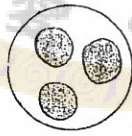
科目：材料科學導論

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- The density of metal niobium is  $8.57 \text{ g/cm}^3$ ; the atomic weight is 92.9. Its lattice parameter is 0.33 nm. Determine whether it has a simple cubic, face-centered cubic, or body-centered cubic crystal structures.
- The metal copper has an FCC crystal structure. If the angle of diffraction ( $2\theta$ ) for the (111) set of planes occurs at  $43.32^\circ$  (first-order reflection) when monochromatic X-rays having a wavelength of 0.154 nm are used, calculate (a) the interplanar spacing of this set of planes, and (b) the lattice parameter of this structure.
- The atomic weight and density for copper at  $1000^\circ\text{C}$  are  $63.5 \text{ g/mol}$  and  $8.4 \text{ g/cm}^3$ , respectively. Calculate the equilibrium number of vacancies per cubic centimeter for copper at  $1000^\circ\text{C}$ . The activation energy for vacancy formation is 0.9 eV/atom. The Boltzmann's constant is  $8.62 \times 10^{-5} \text{ eV/atom-K}$ .
- A plate of iron is exposed to a carbon-rich atmosphere on one side and a carbon-deficient atmosphere on the other side at  $700^\circ\text{C}$ . The carbon concentrations at the two faces are 1.2 and  $0.8 \text{ Kg/m}^3$ , respectively; the thickness of the plate is 5 mm. If the steady-state condition is achieved, calculate the diffusion flux through the plate. Assume the diffusivity of carbon is  $3 \times 10^{-11} \text{ m}^2/\text{s}$  at this temperature.
- The diffusivity of iron in nickel at  $1000^\circ\text{C}$  and  $1200^\circ\text{C}$  are  $9.4 \times 10^{-16}$  and  $2.4 \times 10^{-14}$ , respectively. (a) Determine the values of  $D_0$  and the activation energy (in eV) of diffusion. (b) What is the diffusivity at  $1100^\circ\text{C}$ ?
- The Young's modulus of copper is  $1.1 \times 10^5 \text{ MPa}$ . A piece of copper rod originally 300 mm long is pulled axially in tension with a stress of 300 MPa. If the deformation is entirely elastic, what will be the resultant elongation?

- Using the phase diagram, for a 40 wt% Sn-60 wt% Pb alloy at  $150^\circ\text{C}$ , (a) What phases are present? (b) What is the composition of each phase? (c) What is the mass fraction of each phase?

- The microstructures of three Sn-Pb alloys are:  
(a) (b) (c)



Which points in the phase diagram of problem 7 do the microstructures represent respectively?

- Magnesium oxide has the rock salt crystal structure and a density of  $3.58 \text{ g/cm}^3$ . The atomic weights of magnesium and oxygen are 24.3 and 16, respectively. Determine the unit cell edge length.
- The equilibrium electron and hole concentrations in silicon at room temperature are  $1.33 \times 10^{16}/\text{m}^3$ ; the electron and hole mobilities are 0.14 and  $0.048 \text{ m}^2/\text{V-s}$ , respectively. What is the room temperature electrical conductivity?

試題隨卷繳回