

※注意：請於試卷上「非選擇題作答區」作答，並註明作答之題號。

1. The boiling point of H_2O ($100\text{ }^\circ\text{C}$) is much higher than that of HF ($-83\text{ }^\circ\text{C}$), even though they have similar molecular weights. Explain. (5%)
2. Compare the basicities of pyrrole, pyridine, and piperidine. (5%)
3. The ΔH between *anti* and *gauche* conformation of butane is about 0.9 kcal/mol . Estimate the relative amounts of the two conformers at $25\text{ }^\circ\text{C}$. (5%)
4. An ester $\text{C}_4\text{H}_8\text{O}_2$ yields acid **X** and alcohol **Y** upon hydrolysis. Oxidation of **Y** yields **X**. Identify the ester and show the reactions involved. (8%)
5. Two optically active alkenes, **A** and **B**, have the same molecular formula, $\text{C}_5\text{H}_9\text{Cl}$. After addition of one mole of H_2 to each, **A** is converted to **C** (achiral), and **B** is converted to **D** (optically active). Give the structures of **A**, **B**, **C**, and **D**. (8%)
6. A compound, $\text{C}_{10}\text{H}_{14}\text{O}$, dissolves in NaOH but not in NaHCO_3 . It reacts with aqueous Br_2 to give $\text{C}_{10}\text{H}_{12}\text{Br}_2\text{O}$. The IR spectrum of the compound shows a broad peak at 3250 cm^{-1} and a strong peak at 750 cm^{-1} . The $^1\text{H NMR}$ spectrum shows signals at the following δ values: (1) $\delta = 1.3\text{ ppm}$, *s*, nine H; (2) $\delta = 4.9\text{ ppm}$, *s* (broad), one H; and (3) $\delta = 7.0\text{ ppm}$, *m*, four H. Deduce the structure of the compound. (9%)
7. In one example, treating acetophenone with a peroxy acid converts it into phenyl acetate. Please give the name of this specific reaction and show the detail of reaction mechanism. (10%)
8. With *p*-cresol and aniline, what compound will be obtained from the diazonium coupling reaction? (5%)
9. Phenol is a highly important industrial chemical. Mainly three methods have been used to synthesize phenol commercially. Please give details of these three industrial syntheses (hint: Dow process, alkali fusion, and cumene process). (10%)
10. Cyclopentadiene is so reactive that on standing at room temperature it slowly undergo a reaction with itself. What are the name and the product of this particular reaction? (5%)
11. Which of the following compounds would you expect to undergo aldol self-condensation? Show the product of each successful reaction: (a) trimethylacetaldehyde; (b) cyclobutanone; (c) pentan-3-one; (d) decanal; (e) 3-phenylprop-2-enal. (10%)
12. Please describe three possible synthetic routes of adding a Grignard reagent to a ketone to synthesize a 2-phenylbutan-2-ol. (10%)
13. What experimental approaches are available for determining whether the polymerization of a particular monomer by ionizing radiation proceeds by a radical or ionic mechanism (state at least two approaches and explain)? (10%)

試題隨卷繳回