題號: 290

國立臺灣大學 107 學年度碩士班招生考試試題

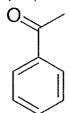
科目: 有機化學(B)

題號:290

節次: 6

共2頁之第1頁

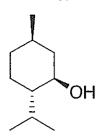
- 1. (10%) Please draw structures for all constitutional isomers with the C<sub>6</sub>H<sub>14</sub> molecular formula.
- 2. (8%) Please draw all resonance structures for the following compound.



3. (12%) Please identify the Lewis acid and the Lewis base on the left side of each reaction.

(b) 
$$\underset{Br}{\overset{Br}{\vdash}}$$
 +  $\underset{Br}{\vdash}$  +  $\underset{Br}{\vdash}$  Br  $\underset{Br}{\overset{\ominus}{\vdash}}$  Br  $\underset{Br}{\overset{Br}{\vdash}}$  Br

4. (8%) Please draw two chair conformations of the following compound and indicate which conformation is lower in energy.



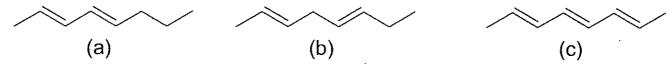
5. (8%) In both <sup>1</sup>H and <sup>13</sup>C NMR spectra, *N*,*N*-dimethylformamide (DMF) shows two separate signals for the methyl groups.

- (a) Please account for the nonequivalence of the methyl groups
- (b) Would you expect the signals to collapse into one signal at high temperature? Give your reason(s).
- 6. (10%) Please deduce the structure of the compound based on the information given below.

Molecular formula: C<sub>4</sub>H<sub>6</sub>O<sub>4</sub>

Infrared-ray spectrum: a broad signal between 2500 and 3600 cm<sup>-1</sup>

- <sup>1</sup>H NMR spectrum: a singlet at 12.1 ppm with a relative integration of 1 and a singlet at 2.4 ppm with a relative integration of 2.
- 7. (8%) Please rank the following compounds in order of increasing the wavelength of maximum absorption (λ<sub>max</sub>).



見背面

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節次:

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6

題號:290

共2頁之第2頁

8. (8%) An acetylide ion can serve as a nucleophile in an S<sub>N</sub>2 reaction:

$$X = leaving group$$
 $H \longrightarrow H \longrightarrow H \longrightarrow R$ 

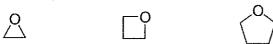
acetylene acetylide ion

Is it possible that this process can be used to synthesize 3,3-dimethyl-1-butyne? Please give your reason(s).

$$=$$

## 3,3-dimethyl-1-butyne

9. (8%) The reactivity of methylmagnesium bromide with ethylene oxide, oxetane, and tetrahydrofuran follows the order of ethylene oxide > oxetane > tetrahydrofuran. Please account for the difference in reactivity.



ethylene oxide

oxetane

tetrahydrofuran

racemic mixture

10.(8%) Treatment of the following optically active alcohol with HBr gives a racemic mixture of alkyl bromide. Please explain the stereochemical results for this reaction.

11.(12%) Please draw structures of major products for the following reactions.

## 試題隨卷繳回