國立臺灣大學113學年度碩士班招生考試試題

科目:有機化學(A).

題號:49 共 7 頁之第 1 頁

節次: 7

第一部份單選題及第二部份多選題考生應作答於『答案卡』

<u>第一部份:單選題</u> (2 pts each)

- There are numerous isomers of hydrocarbon A (C₈H₁₆). Answer Question 1-3.
- 1. How many structural (constitutional) isomers that have a cyclopentane ring do A have? (E) 12
 - (B) 6 (C) 8 (D) 10
- 2. How many constitutional isomers that are straight-chain alkene and have goeometric isomers do A have?
 - (B) 1 (C) 2 (D) 3 (E)4
- 3. For all di-substituted cyclohexane isomers of A, which of the following statement is correct?
 - (A) There are seven structural (constitutional) isomers.
 - (B) Including all (structural + stereo) isomers, five of them are achiral.
 - (C) All trans-isomers are more stable than their corresponding cis-isomers.
 - (D) All of their most stable chair conformations are di-equatorial.
 - (E) Including all (structural + stereo) isomers, four of them are meso compounds.
- Bromobenzene can be prepared from benzene by the following scheme. Answer Question 4-5.

- 4. Which term describes this reaction most properly?
 - (A) elimination (B) addition (C) substitution (D) coupling (E) radical reaction
- 5. How many of the following statement(s) is(are) correct?
 - (a) Fe serves as the catalyst.
 - (b) Bromobenzene is the sole product.
 - (c) The proper molar ratio of benzene, Fe and Br_2 can be 1: 0.1: 1.
 - (d) Benzene serves as the nucleophile.
 - (e) The reaction undergoes the single-step mechanism.
 - **(B)** 1 (C) 2 (D) 3 (E) 4
- Compound A undergoes the following conversion. Answer Question 6-8.

- Which term describes this reaction most properly?
 - (A) ester formation (B) ether formation (C) saponification
 - (D) transamidation (E) transesterification
- 7. When using <u>catalytic</u> amount, which reagent **B** <u>cannot</u> facilitate the conversion?
 - (A) sulfuric acid (B) sodium methoxide (C) sodium hydroxide
 - (D) sodium benzoate (E) sodium hydride.

見背面

國立臺灣大學113學年度碩士班招生考試試題

科目:有機化學(A)

題號:49

共 7 頁之第 2 頁

村日・月機10字(A 節次: 7

8. Which of the following statement is <u>not</u> correct?

- (A) The oxygen O_a in A goes to product C.
- (B) MeOH serves as the nucleophile.
- (C) The reaction will reach equilibrium.
- (D) MeOH serves as the reagent and solvent.
- (E) The reaction undergoes the multi-step mechanism.
- The acyclic form of *D*-glucose undergoes the following reaction to afford the cyclic form. Answer Question 9-10.

9. Which term describes this reaction most properly?

- (A) hydrate formation (B) hemiacetal formation (C) acetal formation
- (D) ether formation (E) ester formation
- 10. As depicted in the cyclic form, which carbon gives the characteristic D notation?

(A) C_a (B) C_b (C) C_c (D) C_d (E) C_e

■ The U.S. Food and Drug Administration (FDA) approved 55 novel therapeutics in 2023. Structures of seven drugs are shown below. Answer Question 11-20.

11. How many chiral centers do these seven drugs have in total?

(A) 7 (B) 8 (C) 9 (D) 10 (E) 11

12. How many functional group(s) shown below cannot be found in these seven drugs?

- (a) ketone, (b) ether, (c) ester, (d) phenol, (e) alcohol, (f) acetal, (g) amine, (h) nitrile, (i) urea, (j) amide
- (A) 1 (B) 2 (C) 3 (D) 4 (E) 5

接次頁

國立臺灣大學113學年度碩士班招生考試試題

科目:有機化學(A)

節次: 7

頁之第 3

13. Several monocyclic or polycyclic aromatic systems can be found in these seven drugs. How many aromatic systems contain 6π electrons in total?

(E) 14

- (A) 6
- (B) 8
- (C) 10
- (D) 12
- 14. How many heterocycles shown below can be found in these seven drugs?
- (a) triazole, (b) pyrimidine, (c) purine, (d) piperidine, (e) morpholine, (f) pyridine, (g) pyridazine

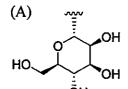
 - (A) 2 (B) 3
- (C) 4
- (D) 5
- 15. Which of the following drug has the most basic function group?
 - (A) ritlecitinib
- (B) quizartinib
- (C) momelotinib
- (D) daprodustat
- (E) repotrectinib
- 16. Which of the following drug has the most acidic function group?
 - (A) ritlecitinib
- (B) quizartinib
- (C) momelotinib
- (D) daprodustat
- (E) repotrectinib

- 17. Which of the following drug has the most acidic C-H?
 - (A) ritlecitinib
- (B) quizartinib
- (C) momelotinib
- (D) daprodustat
- (E) repotrectinib
- 18. How many benzylic hydrogens do these seven drugs have in total?
- (B) 1
- (C) 2
- (D) 3

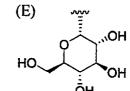
(E) 4

- 19. The cysteine (Cys) residue, a thiol-containing α -amino acid, from target proteins can react with one of
 - these drugs to form a covalent bond, that inhibits the biological activity of proteins. Which drug is it?
- (A) ritlecitinib
- (B) momelotinib
- (C) bexagliflozin
- (D) fruquintinib

20. Bexagliflozin contains a β -C-D-glucoside moiety. Which structure shows the correct stereochemistry?



(B)



- Carbonyl compounds, such as aldehydes, ketones and acid carboxylic acid derivatives, are one of the most important family in organic chemistry. Nine well-known tansformations are shown below. Considering only simple carbonyl componds, answer Question 21-23.
 - (1) Hydrate (gem-diol) formation (2) hemiacetal formation (3) acetal formation,

 - (5) Aldol reaction (6) ester hydrolysis
 - (4) imine formation
 - (7) amide hydrolysis (8) nitrile hydrolysis (9) Claisen condensation.
- 21. How many tansformations are both acid-catalytic (pH<2) and base-catalytic (pH>13)?
- (D) 6
- (E) 7
- 22. How many tansformations are both acid-promoted and base-promoted, and require full equivalent of acids or bases?
 - (A) 1
- (B) 2
- (C) 3

(C) 5

- (D) 4
- (E)5

23. How many tansformations <u>must require</u> aldehydes or ketones as one of starting materials?

- (A) 3
- (B) 4
- (C)5
- (D)6
- (E)7

國立臺灣大學113學年度碩士班招生考試試題

科目:有機化學(A)

題號:49

頁之第 4

節次: 7

24. Which compond in the following reacton is the major oxidant?

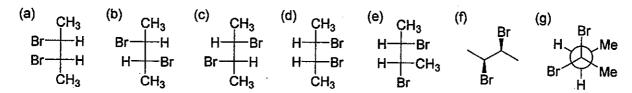
- (A) 2-pentanol
- (B) oxalyl chloride
- (C) DMSO
- (D) TEA
- (E) 2-pentanone
- 25. When acetone is mixed with bleach (sodium hypochlorite), what major product can be produced?
 - (A) 3,3,3-trichloropropanal (B) trichloromethane (C) 1,3-dichloroacetone
- - (D) 2,2-dichloroacetic acid (E) 1,1-dichloroacetone

<u>第二部份:多選題(需全部答對才得分) (2 pts each)</u>

26. Methyl tert-butyl ether (MTBE) can be prepared by the following scheme, where C4 is a four-carbon compound and C1 is a one-carbon compound. What C4, C1 and condition A can be used, respectively?

- (A) isobutene, methanol, sulfuric acid
- (B) potassium tert-butoxide, methanol, DMF
- (C) tert-butanol, methyl iodine, triethylamine/DMF
- (D) tert-butyl chloride, sodium methoxide, DMF
- (E) potassium tert-butoxide, methyl p-toluenesulfonate, DMF

2,3-Dibromobutane is the key model compound to discuss stereochemistry. Answer Question 27-29.



- 27. Which of the following structural representations are used?
- (A) Howard projection (B) Fischer projection (C) Newman projection
- (D) sawhorse projection (E) skeletal line structure
- 28. In the above structural model (a)-(f), which of the following statements are correct?
 - (A) Four structural models are chiral.
 - (B) The model (a) and (g) are identical.
 - (C) The model (b) and (c) are enantiomers.
 - (D) The model (d) and (f) are diastereomers.
 - (E) Three structural models are meso compounds.
- 29. 2,3-Dibromobutane isomers can be prepared from corresponding alkenes with bromine. Which of the following statements are correct?
 - (A) Alkenes serves as the nucleophile.
 - (B) The reaction is through radical addition.
 - (C) The isomer (g) can be prepared from *cis*-2-butene.
 - (D) The isomer (e) can be prepared from trans-2-butene.
 - (E) The reaction undergoes the multi-step mechanism.

接次頁

國立臺灣大學113學年度碩士班招生考試試題

科目:有機化學(A)

節次: 7

題號:49 共 7 頁之第 5 頁

Alkyl halide A-X ($C_5H_{11}X$, X = Cl, Br or I) compounds are common starting materials in synthesis. Some isomers of A-Xs are shown below. Answer Question 30-33.

A1-X

- 30. For all alkyl bromide A-Br (C₅H₁₁Br) compounds, which of the following statements are <u>correct</u>?
 - (A) There are eight structural (constitutional) isomers.
 - (B) Two structural isomers are chiral.
 - (C) Three structural isomers are secondary (2°) bromides.
 - (D) All chiral isomers are secondary (2°) bromides.
 - (E) All chiral carbons are tertiary (3°).
- 31. For all alkyl iodine A-I (C₅H₁₁I) compounds, which of the following statements are incorrect?
 - (A) A3-I is the poorest S_N2 substrates.
 - (B) All primary (1°) A-Is are good S_N2 substrates.
 - (C) A3-I is the best E1 substrate.
 - (D) All A-Is are E2 substrates.
 - (E) A3-I is a poor E2 substrate.
- 32. Both A1-Xs and A2-Xs can be used to prepare elimination products. Which of the following statements are correct?
 - (A) When A1-Xs and A2-Xs are reacted with NaOMe, their major elimination products are the same.
 - (B) When A1-Xs and A2-Xs are reacted with t-BuOK, their major elimination products are the same.
 - (C) When reacted with NaOMe, A1-Xs give a better yield of elimination products than A2-Xs.
 - (D) When reacted with NaOMe, the enantiomers of A2-Xs give the same product.
 - (E) When A1-Xs are reacted with NaOMe, the ranking of reaction rates is A1-I > A1-Br > A1-C1.
- 33. A3-Xs can be used to prepare <u>substitution</u> products. Which of the following statements are <u>correct</u>?
 - (A) When reacted with MeOH, the ranking of reaction rates is A3-I > A3-Br > A3-C1.
 - (B) When reacted with 0.1 M or 0.2 M AcOH, the ranking of reaction rates is 0.1 M < 0.2 M.
 - (C) When reacted with MeOH or NaOMe, the yield of ether formation is MeOH > NaOMe.
 - (D) When reacted with MeOH, A3-Xs give a better yield of substitution products at higher temperature.
 - (E) When reacted with MeOH, the ether product is racemic.
- Compound D can be prepared from benzene through intermediate B. Answer Question 34-48.

- 34. Which reagent A(s) can be used in the above scheme?
 - (A) acetaldehyde
- (B) acetic chloride
- (C) acetic anhydride
- (D) acetic acid
- (E) ethyl choride

- 35. Which reagent C(s) can be used in the above scheme?
 - (A) Zn(Hg), HCl
- (B) H₂H₄, KOH
- (C) NaBH₄, EtOH (D) LiAlH₄, THF
- (E) DIBAL, THF

國立臺灣大學113學年度碩士班招生考試試題

科目:有機化學(A)

題號:49 共7頁之第 6 頁

節次: 7

36. What other synthetic schemes can also be used to prepare B?

- (A) (1) methyl benzoate + MeMgBr, then (2) Jones reagent
- (B) phenylacetylene + HgSO₄/H₂O
- (C) (1) phenylacetylene + excess HBr, then (2) hydrolysis
- (D) (1) ethyl benzene + NBS (1 equiv.) with light, then (2) hydrolysis
- (E) 2-phenyl-1-propene + O₃, then (2) Me₂S

37. What other synthetic schemes can also be used to prepare D?

- (A) (1) benzoic acid + LiAlH₄ in THF, then (2) hydrolysis
- (B) (1) phenylethylene + excess HBr, then (2) hydrolysis
- (C) ethyl benzene + KMnO₄
- (D) styrene + HBr/ROOR with light, then (2) hydrolysis
- (E) (1) ethyl benzene + NBS with light, then (2) hydrolysis
- 38. Which of the following statements are correct?
 - (A) AlCl₃ serves as a catalyst.
 - (B) AlCl₃ serves as a Lewis base.
 - (C) More than one equiv. of AlCl₃ should be added.
 - (D) The IUPAC name of B is 1-phenylethan-1-one.
 - (E) The multi-acetylation is the major concern, yielding a meta-di-acetylation side product.
- Compound **D** can be prepared from cyclohexanol through intermediate **B**. Answer Question 39-41.

- 39. Which synthetic condition A(s) can be used in the above scheme?
 - (A) Lucas' reagent
 - (B) (1) phosphorus tribromide, then (2) sodium acetate
 - (C) concentrated phosphoric acid with heating
 - (D) N,N'-dicyclohexylcarbodiimide (DCC)
 - (E) (1) concentrated hydrobromic acid, then (2) potassium tert-butoxide
- 40. Which reagent C(s) cannot be used in the above scheme?
 - (A) basic KMnO₄ (B) O₃ then Me₂S
- (C) (1) BH₃, then (2) H₂O₂/NaOH
- (D) NaIO₄ (E) MCPBA, then NaOH
- 41. When osmium tetroxide is used as C, which of the following statements are correct?
 - (A) The most stable conformation of \mathbf{B} is the chiar form.
 - (B) Under this condition, the major product **D** has three streoisomers.
 - (C) Under this condition, the major product **D** is racemic.
 - (D) Under this condition, the most stable conformation of **D** is chiar form.
 - (E) Under this condition, there are two major stable conformations of **D**.

接次頁

節次: 7

國立臺灣大學113學年度碩士班招生考試試題

科目:有機化學(A)

題號:49

共 7 頁之第 7 頁

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※ 注意:請於試卷內之「非選擇題作答區」作答,並應註明作答之題號。

第三部份:問答題

42. Use <u>ethanol</u> and <u>toluene</u> as *only* two carbon sources to prepare compound **B**. When designing the synthesis, include <u>ethyl acetoacetate</u> as the key intermediate.

- (A) Provide detailed <u>reagents</u>, <u>schemes</u> and <u>intermediate compounds</u>. (Note: You <u>do not</u> need to provide mechanisms) (6 pts)
- (B) Provide the <u>mechanism</u> of preparation of ethyl acetoacetate from A (C₄H₈O₂). Indicate the <u>driving</u> force of the transformation. (4 pts)
- 43. Amines are important nitrogen-containing organic compounds. Based on numbers of substituents on nitrogen atoms, amines can be primary (1°), secondary (2°) and tertiary (3°). However, preparing one specific type of amines without contamination of other types of amines has posed a great challenge. Using toluene and benzyl amine as the starting material and the target compound, answer the following questions.

- (A) Provide one general method that can prepare <u>any type</u> of amines without contamination of other types of amines. (Note: Use benzyl amine from toluene as an example) (2 pts)
- (B) Provide three specific methods that <u>only</u> can synthesize primary (1°) amines. (Note: Use benzyl amine from toluene as an example) (6 pts)

試題隨卷繳回