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### 國立臺灣大學 111 學年度碩士班招生考試試題

科目: 有機化學(A)

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(A) I

(A) I

(A) I

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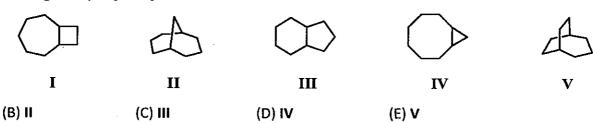
#### 注意:本試題包含單選題及問答題兩部份

Part I. 單選題 (25 題,共50分)※ 注意:請於試卷內之「選擇題作答區」依序作答。

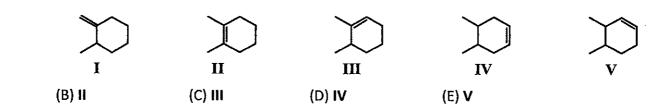
- 1. Which compound would be expected to have the lowest  $pK_a$ ?
- (A) 4,4-dichlorobutanoic acid
- (B) 3,4-dichlorobutanoic acid
- (C) 3,3-dichlorobutanoic acid

- (D) 2,3-dichlorobutanoic acid
- (E) 2,2-dichlorobutanoic acid
- 2. Which of the following amino acids is theoretically capable of existing in diastereomeric forms?

3. Which of the following is bicyclo[6.1.0]nonane?



4. Which molecule would have the lowest heat of hydrogenation?



5. Select the structure of the major product in the following reaction.

- (A) cyclopentanol
- (B) 1-cyclopentylcyclopentane
- (C) cyclopentanone

- (D) cyclopentene
- (E) 1-(1-hydroxycyclopentyl)-1-hydroxycyclopentane
- 6. Select the correct reagent(s) for the following reaction:

- (A) LiAlH<sub>4</sub>/ether; then H<sub>3</sub>O<sup>+</sup>
- (B) NaBH<sub>4</sub>; then H<sub>3</sub>O<sup>+</sup>
- (C) H<sub>2</sub> with Pt/C

- (D) Two of these choices.
- (E) All of these choices.
- 7. Which of the compounds listed below is more acidic than 1-butanol?
- (A) Diethylmalonate
- (B) 2-Butanone

(C) Ethyl pentanoate

- (D) Two of these choices.
- (E) All of these choices.

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8. How many isomers, including stereoisomers, exist for the triacylglycerol which, on saponification, gives glycerol, 2 molar equivalents of palmitate and 1 molar equivalent of stearate?

- (A) 2
- (B)3
- (C)4
- (D) 5
- (E) 6

9. Which of the following would **NOT** undergo racemization in base?

- (A) (R)-3-methyl-4-heptanone
- (B) (R)-3-methyl-2-heptanone
- (C) (R)-4-methyl-2-heptanone

- (D) (R)-2,4-dimethyl-3-heptanone (E) All of these choices will undergo racemization in base.

10. Why is CH<sub>3</sub>ONa not used in the Claisen condensation of ethyl acetate?

- (A) CH<sub>3</sub>O<sup>−</sup> is a weaker base than the CH<sub>3</sub>CH<sub>2</sub>O<sup>−</sup> which is used.
- (B) CH<sub>3</sub>O<sup>-</sup>Na<sup>+</sup> is more difficult to prepare than CH<sub>3</sub>CH<sub>2</sub>O<sup>-</sup>Na<sup>+</sup>.
- (C) CH<sub>3</sub>O<sup>−</sup> would abstract a proton from the ethyl group of the ester.
- (D) Use of CH<sub>3</sub>O<sup>-</sup>Na<sup>+</sup> would result in transesterification.
- (E) CH<sub>3</sub>O<sup>-</sup>Na<sup>+</sup> can be used as well as CH<sub>3</sub>CH<sub>2</sub>O<sup>-</sup>Na<sup>+</sup>.

11. A glycoside is a compound which contains the structural features of these classes of organic compounds:

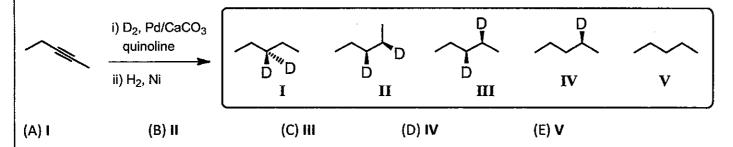
- (A) Aldehydes and alcohols.
- (B) Acetals and alcohols.
- (C) Hemiacetals and alcohols.

- (D) Ketones and alcohols.
- (E) Alcohols and carboxylic acids.

12. Which compound would show optical activity?

- (A) trans-1,4-Dimethylcyclohexane
- (B) cis-1,4-Dimethylcyclohexane
- (C) trans-1,4-Dimethylcycloheptane
- (D) cis-1,4-Dimethylcycloheptane
- (E) More than one of these choices.

13. Which would be the *major* product of the following reaction sequence?



14. In the presence of light, ethane (1 mol) reacts with chlorine (1 mol) to form which product(s)?

- (A) CH2CICHCI2
- (B) CH<sub>3</sub>CHCl<sub>2</sub>

(C) CH<sub>3</sub>CH<sub>2</sub>Cl

- (D) CICH2CH2CI
- (E) All of these choices.

15. Which reagent would serve as the basis for a simple chemical test that would distinguish between the pair of compounds listed below?



- (A) AgNO<sub>3</sub> in H<sub>2</sub>O
- (B) Dilute NaOH
- (C) Dilute NaHCO<sub>3</sub>

- (D) C<sub>6</sub>H<sub>5</sub>SO<sub>2</sub>Cl/OH<sup>-</sup>, then H<sub>3</sub>O<sup>+</sup>
- (E) Dilute HCl

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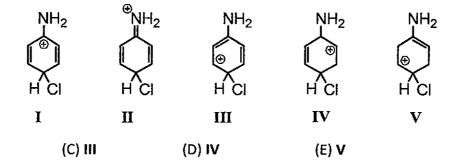
16. What would be the final organic product of the following reaction?

Br 
$$\stackrel{\text{i) Mg, ether}}{\underset{\text{ii) CO}_2}{\longrightarrow}} \stackrel{\text{i) LiAlH}_4, \text{ ether}}{\underset{\text{ii) H}_3\text{O}^+}{\longrightarrow}} ?$$

- (A) (CH<sub>3</sub>)<sub>3</sub>CCO<sub>2</sub>H
- (B) (CH<sub>3</sub>)<sub>3</sub>CCOCH<sub>3</sub>
- (C) (CH<sub>3</sub>)<sub>3</sub>CCH<sub>2</sub>OH

- (D) (CH<sub>3</sub>)<sub>3</sub>COCH<sub>3</sub>
- (E) (CH<sub>3</sub>)<sub>3</sub>CCO<sub>2</sub>CH<sub>3</sub>

17. Which of the following contributors to the resonance stabilized hybrid formed when aniline undergoes *para*-chlorination would be exceptionally stable?



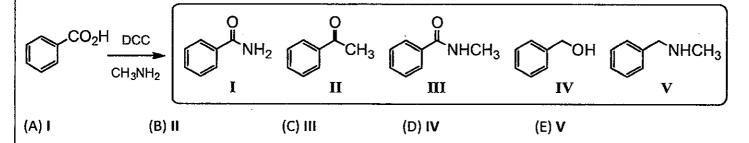
- (A) I
- (B) II

18. Which would be the best method for converting 3,3-dimethyl-1-pentene into 3,3-dimethyl-2-pentanol?

- (A) Hg(OAc)<sub>2</sub>/THF-H<sub>2</sub>O; then NaBH<sub>4</sub>,OH<sup>-</sup>
- (B) H<sub>3</sub>O<sup>+</sup>, heat
- (C) concd. H<sub>2</sub>SO<sub>4</sub>; then H<sub>2</sub>O, heat
- (D) BH<sub>3</sub>:THF; then H<sub>2</sub>O<sub>2</sub>, OH<sup>-</sup>

(E) HBr; then NaOH/H2O

19. What would be the final organic product of the following reaction?



20. Which is the only one of these reagents which will react with benzene under the specified conditions?

(A) Cl<sub>2</sub>, FeCl<sub>3</sub>, heat

(B) H<sub>2</sub>, 25 °C

(C) Br<sub>2</sub>/CCl<sub>4</sub>, 25 °C, dark

(D) KMnO<sub>4</sub>/H<sub>2</sub>O, 25 °C

(E) H<sub>3</sub>O<sup>+</sup>, heat

21. The overall conversion RBr → RCH<sub>2</sub>NH<sub>2</sub> can be accomplished by successive application of which of these sets of reagents?

- (A) Mg, ether; then NH<sub>3</sub>
- (B) NaN<sub>3</sub>; then LiAlH<sub>4</sub>, ether
- (C) NaCN; then LiAlH<sub>4</sub>, ether

- (D) H<sub>2</sub>C=O; then NH<sub>3</sub>
- (E) H<sub>2</sub>NOH; then LiAlH<sub>4</sub>, ether

22. Which reagent(s) could be used to carry out the following transformation?



- (A) Zn(Hg), HCl, reflux
- (B) LiAlH<sub>4</sub>, ether
- (C) HSCH<sub>2</sub>CH<sub>2</sub>SH, BF<sub>3</sub>; then Raney Ni

- (D) All of these choices.
- (E) Two of these choices.

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23. Which sequence of reactions would be utilized to achieve the following conversion?

- (A) 2 CH<sub>3</sub>MgBr, then NH<sub>4</sub><sup>+</sup>
- (B) HOCH2CH2OH, H3O+; LiAlH4, ether; 2 CH3MgBr, then NH4+
- (C) HOCH2CH2OH, H3O+; CH3CH2MgBr, then NH4+
- (D) HOCH<sub>2</sub>CH<sub>2</sub>OH, H<sub>3</sub>O<sup>+</sup>; H<sub>2</sub>, Pt; CH<sub>3</sub>OH, NH<sub>4</sub><sup>+</sup>
- (E) None of these choices.
- 24. The preferred conformation of cis-3-tert-butyl-1-methylcyclohexane is the one in which:
- (A) the tert-butyl group is axial and the methyl group is equatorial.
- (B) the methyl group is axial and the tert-butyl group is equatorial.
- (C) both groups are axial
- (D) both groups are equatorial.
- (E) the molecule exists in a boat conformation.
- 25. What is the chief product of the Friedel-Crafts alkylation of benzene with 1-butene and HF?
- (A) 2-phenylbutane
- (B) butylbenzene
- (C) 2-methyl-1-phenylpropane

- (D) t-butylbenzene
- (E) 2,2-diphenylbutane

## Part II. 問答題 (5題,共50分) ※ 注意:請於試卷內之「非選擇題作答區」標明題號依序作答。

- 1. Please describe the following name reactions; (a) Jones oxidation (4 points), (b) Wittig reaction (4 points), and (c) Clemmensen reduction (4 points).
- 2. Although pyridine and pyrrole are structurally similar, they show a huge difference in their basicity. The  $pK_a$  values of the corresponding aminium ions are given below. Please indicate which one is a stronger base and provide explanation(s) for the discrepancy. (8 points)

pyridine



pKa of corresponding aminium ion

- 3. What is Williamson ether synthesis? Give a specific example and provide a step-by-step mechanism to account for the conversion. (10 points)
- 4. What is Fischer esterification? Give a specific example and provide a step-by-step mechanism to account for the conversion.
- 5. Provide structures for the following groups; (a) benzyl, (b) benzoyl, (c) methanesulfonyl, (d) isopropyl, and (e) ethoxy. (10 points)

# 試題隨卷繳回