國立臺灣大學103學年度轉學生招生考試試題

題號: 50

科目: 有機化學

題號: 50

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Problem 1. (18 pts)

For each of the following pairs, determine if the compounds are (a) constitutional isomers, (b) enantiomers, (c) diastereoisomers, (d) configurational (or geometrical isomers, (e) conformers, or (f) identical molecules.

$$^{(iv)}$$
 CHO CHO
 $^{+0}$ HO $^{-1}$ H $^{+-}$ OH
 $^{-1}$ CH $_2$ OH CH $_2$ OH

Problem 2. (4 pts)

An unknown compound is optically active and has the molecular formula of C_5H_{10} . Draw all possible isomers of the compound.

Problem 3. (6 pts)

What is the IUPAC name of the following compound? (R)- or (S)- designation and (E)- or (Z)- configuration have to be included in the nomenclature.

見背面

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Problem 4. (6 pts)

Assign the PK_a values (2.18, 8.95, 10.53) to lysine.

Problem 5. (9 pts)

Tell whether each of the following reactions is likely to be S_N1 , S_N2 , E1, E1CB, or E2.

Problem 6. (21 pts)

Identify the reagents and products A-G in the following reaction scheme.

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Problem 7. (8 pts)

(a) Propose a structure for the compound of molecular formula $C_4H_{10}O$ whose proton NMR and IR data are summarized as follows: (5 pts)

 ^{1}H NMR: δ 3.41 (doublet, 2H), 2.42 (singlet, 1H), 1.77 (nonet, 1H), 0.91 (doublet, 6H)

IR: ~3500 cm⁻¹ (broad)

(b) Use tree diagram to explain the ¹H NMR splitting. (3 pts)

Problem 8 (18 pts)

Predict the products of the following reactions.

$$\begin{array}{c|c} \text{(d)} & \text{CH}_3 & \text{HNO}_3 \\ \hline & \text{H}_2\text{SO}_4 \\ \end{array}$$

(f)
$$C_6H_5$$
 C_6H_5 C_6H_5

Problem 9 (5 pts)

Provide reaction mechanisms for the Edman degradation, an effective way for shortening a peptide chain.

$$H_2N$$
 Peptide (2) HCI (3) (2) (3) (4) (4) (4) (4) (5) (5) (4) (5) (4) (5) (4) (5) (4) (5) (5) (4) (5) (4) (5) (5) (5) (6) (7) (7) (7) (8) (9) (9) (1) (1) (1) (1) (2) (3) (4) (4) (4) (4) (5) (5) (5) (6) (7) (7) (7) (7) (8) (7) (8) (9) (9) (9) (1)

Problem 10 (5 pts)

Provide reaction mechanisms for aldol condensation.

$$H_3C$$
 H H_3C H H^+

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