類號: 60

國立臺灣大學106學年度碩士班招生考試試題

科目:有機化學(A)

節次: 7

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第一部份: 單遇題 (3pts each) 請用 2B 鉛筆作答於答案卡,並先詳閱答案卡上之「畫記說明」。

- 1. How many of the following species can primarily serve as a nucleophile?
 - (a) proton (b) hydroxide ion (e) triethylamine (c) boron trifluoride (d) triphenylphosphine
 - (f) methanol (g) pyridine
 - (A) 3
- (B) 4

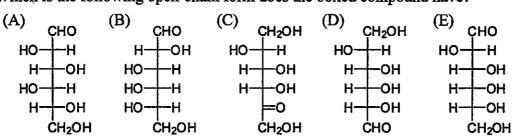
(B)2

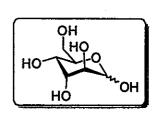
- (C) 5
- (D) 6 (E) 7
- 2. How many of the following reactions can afford the major product as shown?

- (A) 1
- (C) 3
- (D) 4
 - 4 (E) 5
- (e) $CF_3 \xrightarrow{HNO_3} O_{2N}$ CF_3
- 3. Which is the major product of the following synthetic scheme?

$$(A) \underset{N_3}{\overset{CH_3}{\longleftrightarrow}} CI \qquad (B) \underset{N_3}{\overset{N_3}{\longleftrightarrow}} (C)$$

4. Which is the following open-chain form does the boxed compound have?





5. Which of the following transformation(s) can yield acetophenone (C₈H₈O) as the major product?

(a) sodium acetylide
$$\frac{\text{HgSO}_4}{\text{DMF}}$$
 $\frac{\text{HgSO}_4}{\text{H_2SO}_4, \text{H}_2O}$ $\frac{\text{(b)}}{\text{benzene}}$ $\frac{\text{Ac}_2O}{\text{DMSO}}$ $\frac{\text{(c)}}{\text{styrene}}$ $\frac{\text{H}_2\text{SO}_4}{\text{H}_2O}$ $\frac{\text{(1) LiAlH}_4}{\text{(2) H}^+, \text{H}_2O}$

(d) (e) (1) NaNO₂ ethylbenzene
$$\frac{\text{NBS (2eq)}}{\text{light}} \stackrel{\text{OH}^{+}, \text{ H}_2\text{O}}{\text{OH}^{-}, \text{ H}_2\text{O}} = \text{aniline} \frac{\text{H}_2\text{SO}_4, 5^{\circ}\text{C}}{\text{(2) CuCN}} = \frac{\text{(1) MeMgBr}}{\text{(2) H}^{+}, \text{ H}_2\text{O}}$$

- (A) de (B) abe
- (C) ade
- (D) bde
- (E) abcde
- 6. Which of the following nitrogen-containing compound is most basic?
 - (A) acetamide
 - (B) ethylamine
- (C) pyridine (D) aniline
- (E) imidazole

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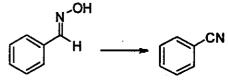
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7. How many of the following reagents can complete the transformation?

- (a) MnO_2
- (b) Ac_2O

(C) 3

- (c) DCC (d) NaOEt
- (e) dilute H₂SO_{4(aq)} (f) NaBH₄
- (A) 1
- (B) 2
- (D) 4



8. In the reaction of sodium acetate (0.1M) and (R)-3-bromo-3-methylhexane (0.1M) in EtOH, which of the following statement is correct? ([S] = concentration of S)

- (A) The reaction is faster when [sodium acetate] is 0.2M.
- (B) The reaction yields a racemic mixture. Therefore the products are achiral.
- (C) The major product is 3-ethoxyl-3-methylhexane
- (D) The major product has the S configuration.
- (E) The reaction is faster in chloroform than in EtOH.

9. How many of the following species are aromatic?





(A)3

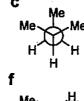
- (B) 4
- (C)5
- (D)6
- (E)7

The representative conformations (a to f) of 2-methylbutane and its potential energy diagram of internal rotation (position 1 to 6) are shown here. Answer Question 10-12.

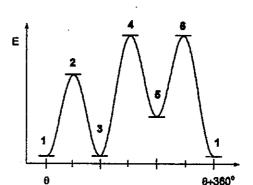












10. In the energy diagram, what is the conformation in the position 1?

- (A) a
- (B) **b**
- (C) c
- (D) **d**

11. Starting from the position 1 on the left side, what is the rotational sequence of conformations to the next position 1 on the right side?

- (A) aecdbfa
- (B) bfcdaeb
- (C) cdbfaec
- (D) bfaecdb
- (E) adbecfa

12. Which of the following statement is correct?

- (A) The position 2, 4 and 6 are stable conformations.
- (B) When temperature is rising, there will have more molecules in the position 2 than in the position 5.
- (C) 2-Methylbutane is achiral; therefore, conformation a to f are also achiral.
- (D) The conformations in the position 1 and 3 are enantiomers.
- (E) The conformation d is unstable due to gauche interactions.

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13. How many of the following reagents can complete the transformation?

- (a) 2-methylpropan-2-ol + BF₃
- (b) 2-methylpropene + H₃PO₄
- (c) 1-chloro-2-methylpropane + AlCl₃ (d) t-butyl chloride + FeCl₃
- (B) 1 (C) 2 (D) 3

14. Which compound most likely has the following NMR data?

¹H NMR (CDCl₃): δ 2.56 (s, 3H), 3.88 (s, 3H), 6.94 (d, J = 8.5Hz, 2H), 7.95 (d, J = 8.5Hz, 2H)

¹³C NMR (CDCl₃): δ 21.3, 26.6, 125.6, 128.4, 128.7, 133.8, 137.2, 138.3, 198.3

(A) (B) (C)

15. In the following five pairs of ten compounds, which of the statement is correct?

(a)

- (A) There are three pairs of enantiomers.
- (B) There are three pairs of diastereomers
- (C) There are three meso compounds.
- (D) There are five chiral compounds.
- (E) There are two pairs of constitutional isomers.

16. Which is the following reaction does NOT give alcohol as the major product?

(D)

17. When reacting with hept-6-en-2-one (C₇H₁₂O), how many of the following reagents will show "positive"

- (a) Jones reagent (b) iodoform test (c) Lucas' reagent (d) KMnO_{4(aq)}
- (e) 2,4-dinitrophenylhydrazine test (f) Br₂/CCl₄ (g) Tollens' reagent
- (A)3(B) 4 (C)5(D) 6(E) 7

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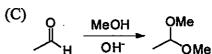
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18. Which of the following compound contain the most acidic C-H bond?

- (A) 1-hexyne
- (B) 2,4-hexadiene
- (C) 2-hexanone
- (D) 3,4-hexanedione
- (E) 2,4-hexanedione

19. Which of the following transformation(s) can yield the major product as shown?

(B) MeLi



第二部份:問答題

- 20. Compound B can be prepared through compound A by the following synthetic plan:
 - (A) Provide the IUPAC name of compound A. (3pts)
 - (B) Design a synthetic scheme to prepare A using acetylene (C₂H₂), isopropanol, isobutanol as sole organic materials along with any other inorganic materials and common organic solvents. (8pts)
 - (C) Design a synthetic route to prepare B from compound A. (6pts)

21. Cyclopentanone can be prepared from dimethyl succinate through compound C by the following synthetic scheme. Compound C is an "achiral" dimethyl ester and does NOT contain any chiral centers. Provide suitable reagents to complete the synthetic scheme. (8pts)

- 22. Design a synthetic scheme to prepare ethyl 3-bromobenzoate from benzene. (You may use any other required reagents. If some of your reactions produce two major products, you can separate them.) (6pts)
- 23. Give a detailed reaction mechanism for the following reactions.

$$(A) \bigcirc OH + \bigcirc O \xrightarrow{H^+} \bigcirc OOO$$
 (4pts)

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