國立臺灣大學100學年度碩士班招生考試試題

科目:有機無機

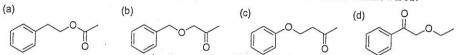
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有機化學共50分

(一) 單選題(每題2分,共30分,請依題號順序於『選擇題作答區』內作答。)

(1) The 1 H NMR spectrum of an unknown compound A ($C_{10}H_{12}O_{2}$) shows: δ 7.2-7.4 (5H, multiplet), 4.28 (2H, triplet), 2.92 (2H, triplet), 2.0 (3H, singlet). What is the structure of the unknown A?



(2) How many of the following three addition reactions would favor the syn-adduct?

(3) Which of the following reactions would give an alkene as the major product?

(4) How many of the following four compounds are aromatic?

$$\bigoplus_{\mathbb{R}} \mathbb{R}$$

$$\mathbb{R}$$

(5) Which of the following reactions would not be promoted under basic conditions?

(a)
$$CI$$
 CEN H_2O NH_2 (b) CEN H_2O NH_2

(6) Which position would be the most reactive towards electrophilic aromatic substitution?

(7) How many of the following eight substituents are activating, ortho-para directing substituents?

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(9) Which of the following chemical equilibriums would have the equilibrium constant K < 1?

(a) (b) (b)
$$H + H_2O = HOOH (Cl_3C) + HOOH (Cl_3C) + HOOH (HOOH) (HOOH$$

(10) Which of the following reactions would give the highest heat of hydrogenation (the heat energy released)?

(a)
$$H_2$$
 (b) H_2 $Cat.$ (c) H_2 $Cat.$ (d) H_2 $Cat.$

(11) How many of the following compounds are chiral?

$$B_r$$
 B_r B_r

(12) How many stereogenic centers of S configuration are in the following compound?

(13) How many of the following three expressions are incorrect?

(14) How many of the following three reactions allow one to obtain the target epoxide compound?

(15) How many of the following three reactions allow one to obtain the target alcohol compound?

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(二) 問答題(共20分,請於『非選擇題作答區』內作答。)

(16) Name the functional group (每一小題 1 分, 共 4 分)

(17) Propose the mechanisms for the following reactions. (每一小題 2 分,共 8 分)

(a)
$$O + Ph_3P = CH_2$$
 CH_2

(b) $O + OH$
 OCH_3

(c) $KMnO_4$
 OH
 OH

(18) Give the structures of intermediates or products A through H for each of the following reactions:

(a)
$$\frac{\text{(i) BH}_3 \cdot \text{THF}}{\text{(ii) H}_2\text{O}_2, \text{OH}^-} \text{ A (IR: 3500 cm}^{-1}) \xrightarrow{\text{PCC}} \text{ B (IR:~1725 cm}^{-1}) \xrightarrow{\text{Li}} \text{ C}$$

(c)
$$H_{3}$$
C $KOH, Bu_{4}NOH$ $E(C_{8}H_{10}O)$ $Me_{2}CuLi$ $F(C_{9}H_{14}O)$ (IR:~1710 cm⁻¹)

(d)
$$G$$
 G Zn, HCl H

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無機化學共50分

1 In the reaction of ethylene hydrogenation with $RhH_2Cl(PPh_3)_3$, a rate law as below is obtained.

 $-d[RhH_2Cl(PPh_3)_3]/dt = k_{obsd}[RhH_2Cl(PPh_3)_3]$

 $k_{obsd}^{-1} = A([PPh_3]/[C_2H_4]) + B$

- (a) Propose a mechanism that can explain the rate law. (8 %)
- (b) What results you would expect if RhH₂Cl(PPh₃)₃ and RhD₂Cl(PPh₃)₃ in 1:1 ratio were used for the reaction. Explain. (8 分)
- 2 CH₄ is insoluble in water; H₂S however, is acidic in water; whereas NaH reacts with water vigorously by evolving H₂.
 - (a) Characterize the hydrogen in each of these three compounds. (8 分)
 - (b) Why does the H element behave so differently in these three compounds? (8 分)
- 3 The reaction of fac-Mn(Me)(PPh₃)₂(CO)₃ and ¹³CO has been studied by IR.
 - (a) Draw the structure for the organometallic reactant and product(s). Assign their point groups. Give their IUPAC names. (8分)
 - (b) Each product of this reaction has a new, rather strong IR band that is distinctly different energy from any band in the reactant. Account for this band with their symmetry and bonding characters, and predict its approximate location in cm⁻¹ in the IR spectrum. (10分)

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